

Plastic Electronics

Joaquim Puigdollers

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Origins...

Since the discovery that the conductivity of *trans*-polyacetylene increases seven orders of magnitude upon oxidation with iodine ...

J.C.S. CHEM. COMM., 1977, 578-580

Synthesis of Electrically Conducting Organic Polymers: Halogen Derivatives of Polyacetylene, $(CH)_x$

By HIDEKI SHIRAKAWA, EDWIN J. LOUIS, ALAN G. MACDIARMID,* CHWAN K. CHIANG, and ALAN J. HEEGER
(Department of Chemistry and Department of Physics, Laboratory for Research on the Structure of Matter, University of Pennsylvania, Philadelphia 19104)

Summary When silvery films of the semiconducting polymer, *trans* 'polyacetylene', $(CH)_x$ are exposed to chlorine, bromine, or iodine vapour, uptake of halogen occurs, and the conductivity increases markedly (over seven orders of magnitude in the case of iodine) to give, depending on the extent of halogenation, silvery or silvery-black films, some of which have a remarkably high conductivity at room temperature.



A.J. Heeger



A.G. MacDiarmid



H. Shirakawa

Nobel Prize Chemistry 2000

Technological Interest

Wired 8.07: Molecular Electronics Will Change Everything - Microsoft Internet Explorer

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Issue 8.07 | July 2005

Molecular Electronics Will Change Everything

Pg 1 of 3 [»](#)

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The Next Big Thing is very, very small. Picture trillions of transistors, processors so fast their speed is measured in terahertz, infinite capacity, zero cost. It's the dawn of a new technological revolution - and the death of silicon. Can you say Thiophene Ethynylene Valley?

By Rick Overton

Once again, Jim Tour has forgotten to breathe. Sitting in his office at Rice University in Houston, he's telling the story of how he was heckled while giving a speech at the 1995 Marvel Symposium in Tucson, Arizona - an event that attracts the world's foremost chemists - and it's making him even more animated

WIRED BLOGS

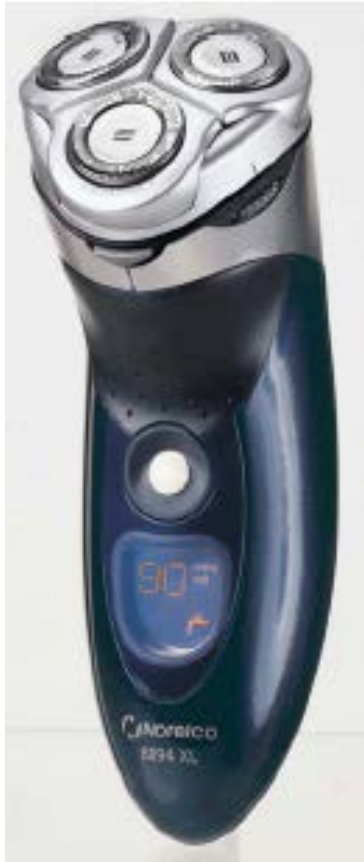
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First products..

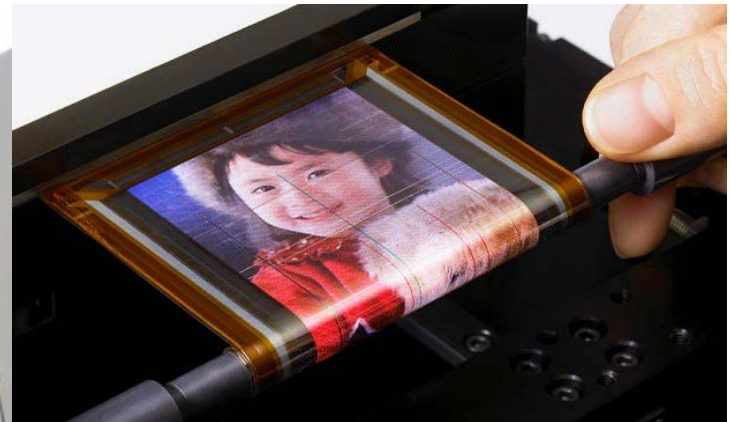


PHILIPS

MONOCROMATIC



Today...Future



Technological interest

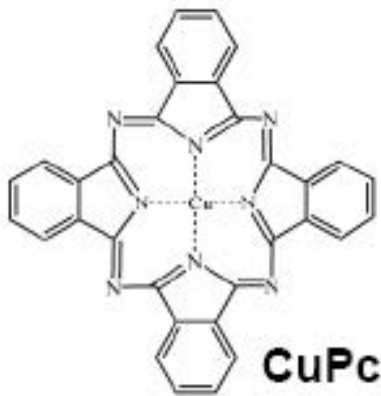


Low processing T (< 200°C)
Flexible
**Huge variety of polymers /
organic molecules**

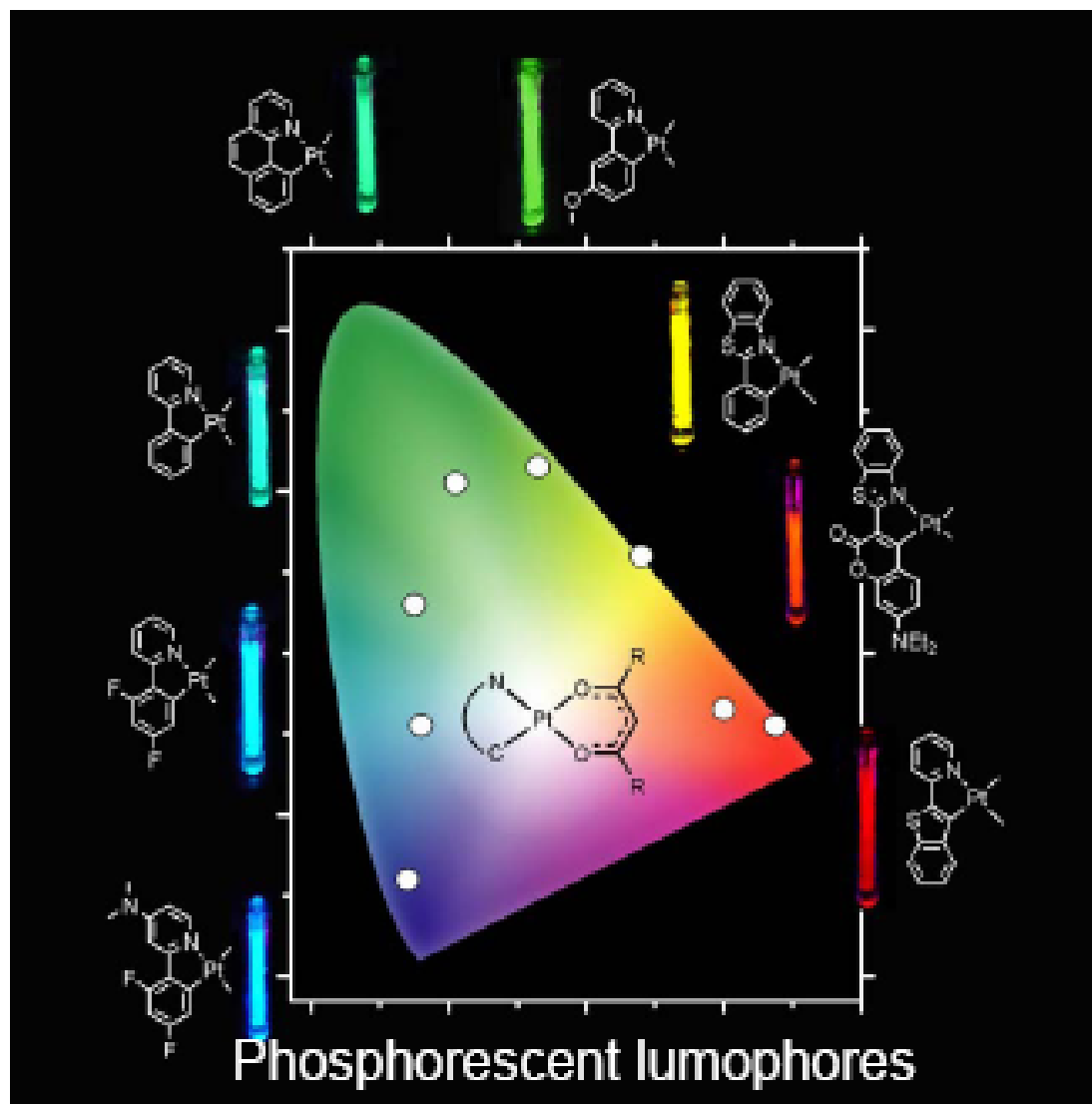


Dyes

Pigments: stables, small-molecule, visible optical absorption

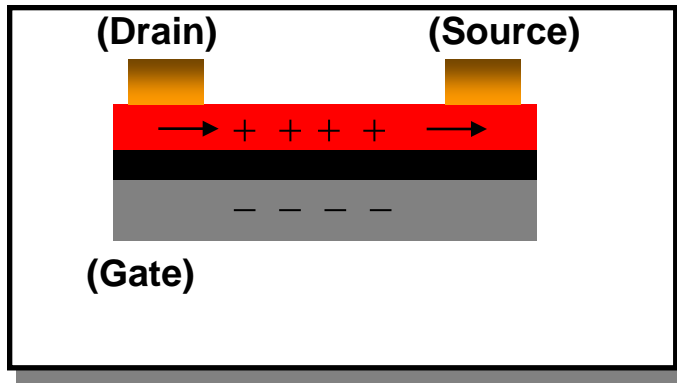


Molecules with adjustable properties

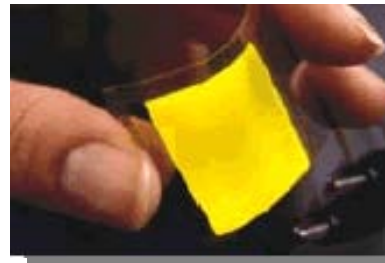


Organic Devices

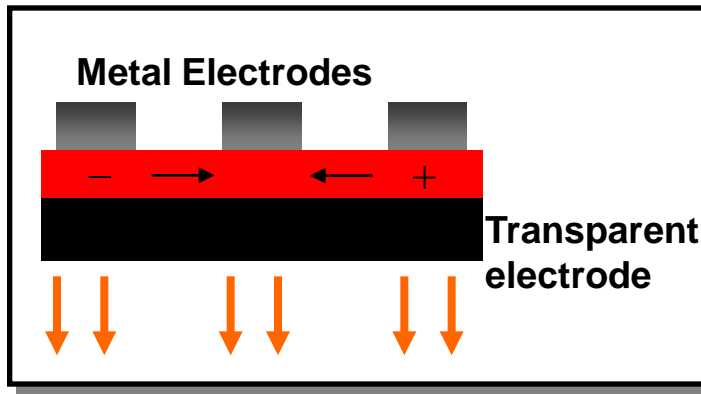
Thin-film transistor (TFT)



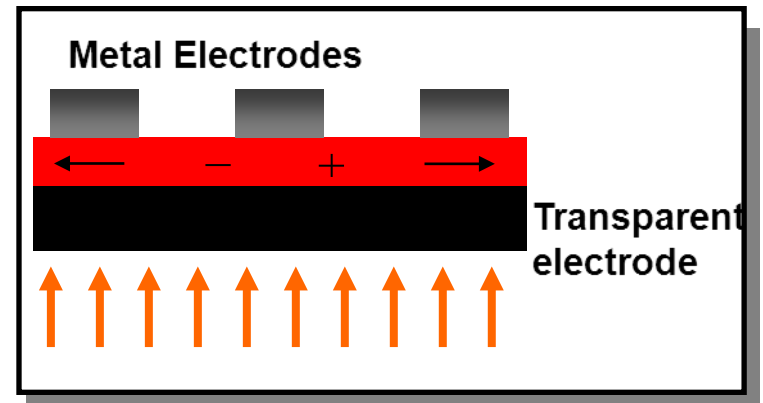
Thin-Film



Light emitter diode (OLED)

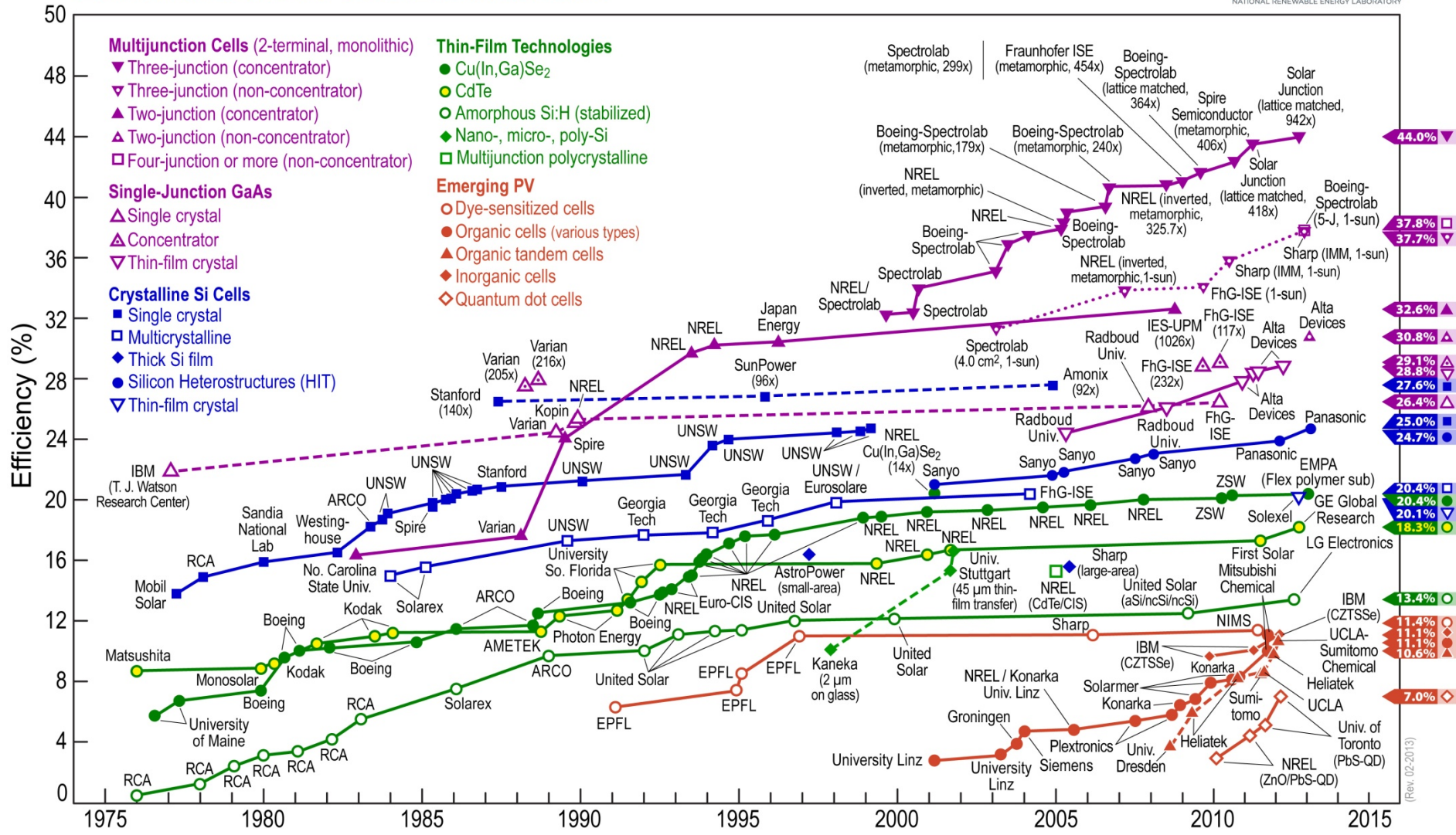


Solar cell

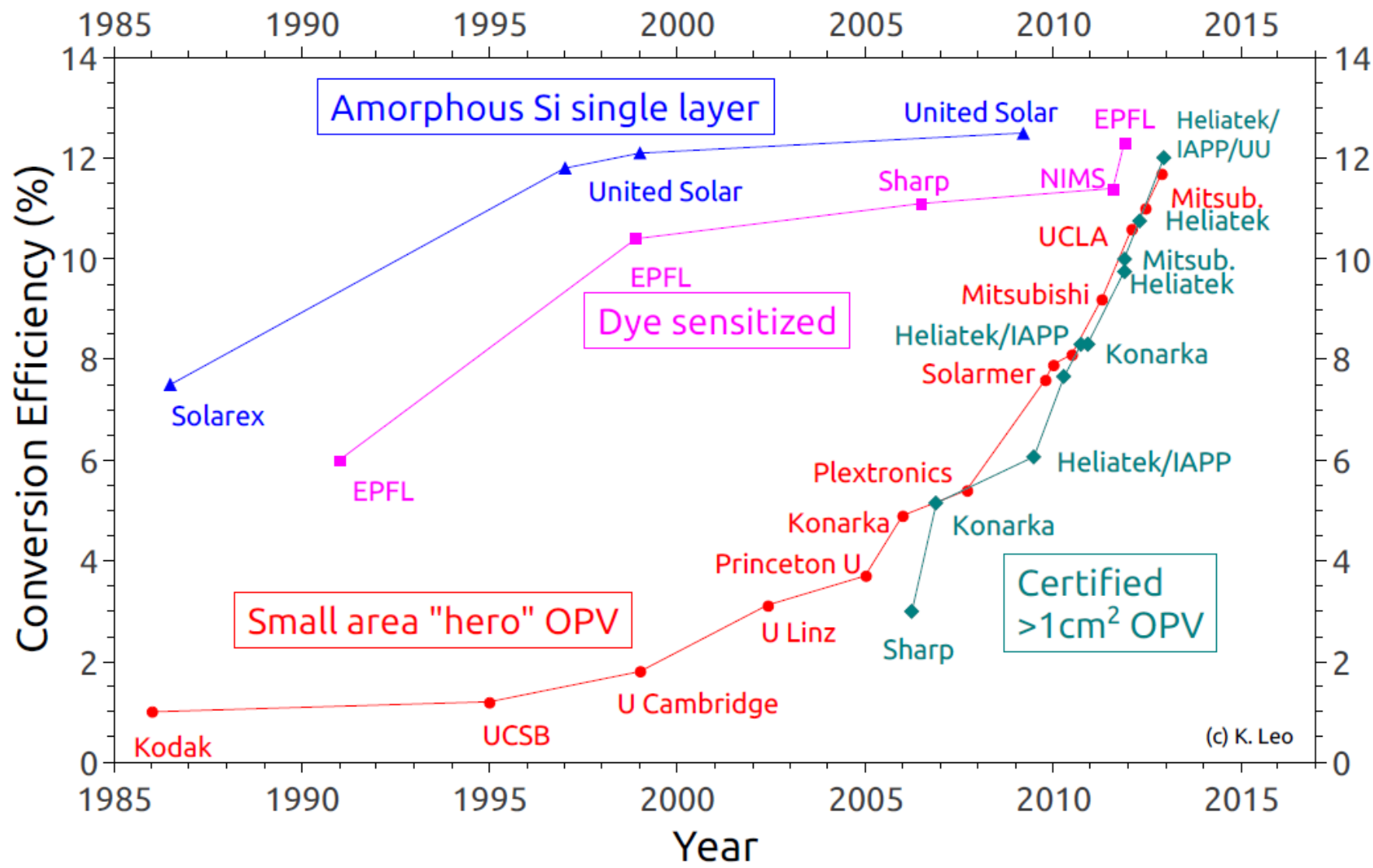


Organic Solar Cells

Best Research-Cell Efficiencies



(Rev. 02-2013)

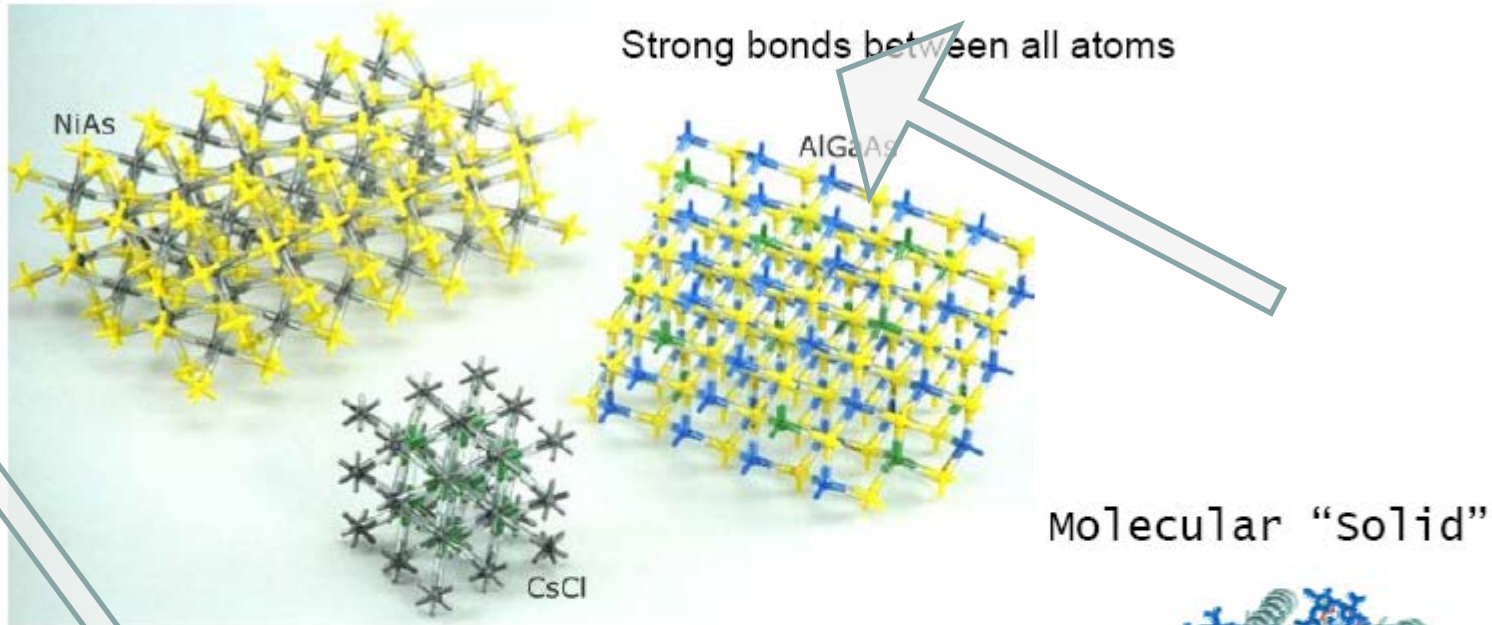


(c) K. Leo

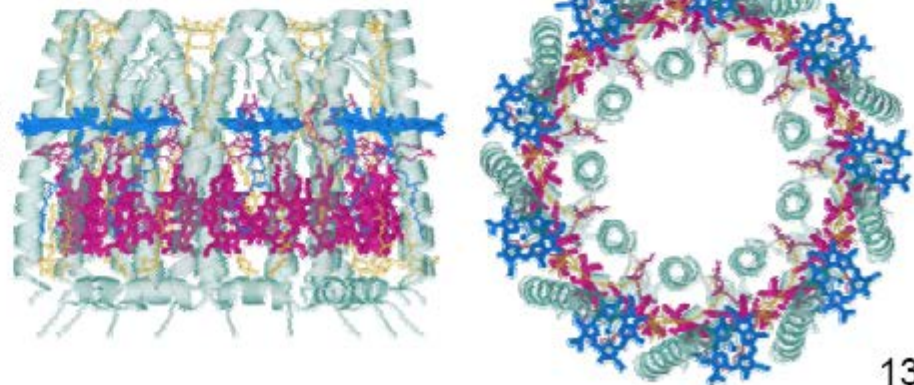
Organic material
(or molecular material) ?

Organic material (or molecular material) ?

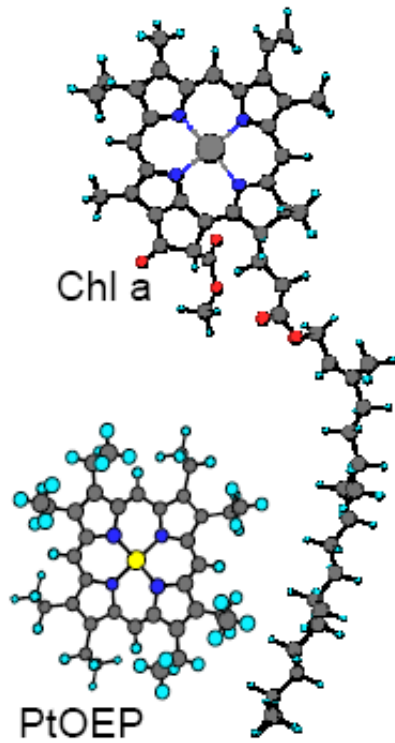
Covalent, Ionic and Metallic Solids



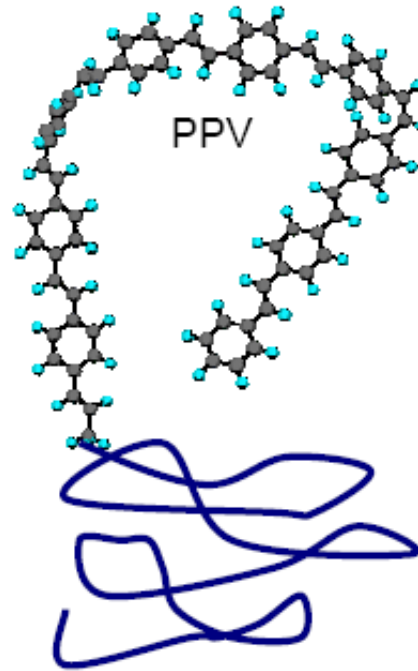
Aggregates of atoms (molecules) that are strongly bonded are held together via weaker intermolecular forces



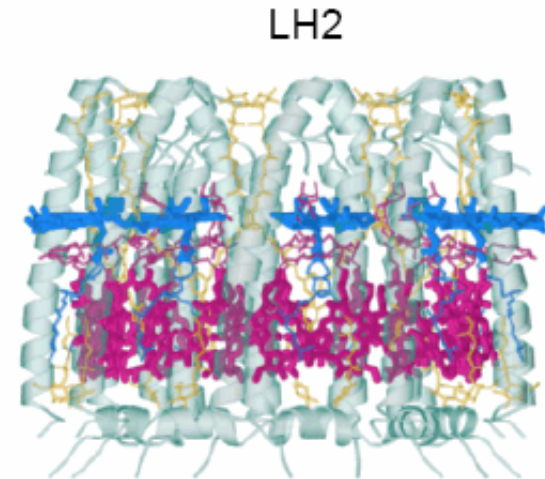
Polymer vs small molecule



Monomers



Polymers



Biological Molecules



Complexity

Organic Solar Cells

Two approach

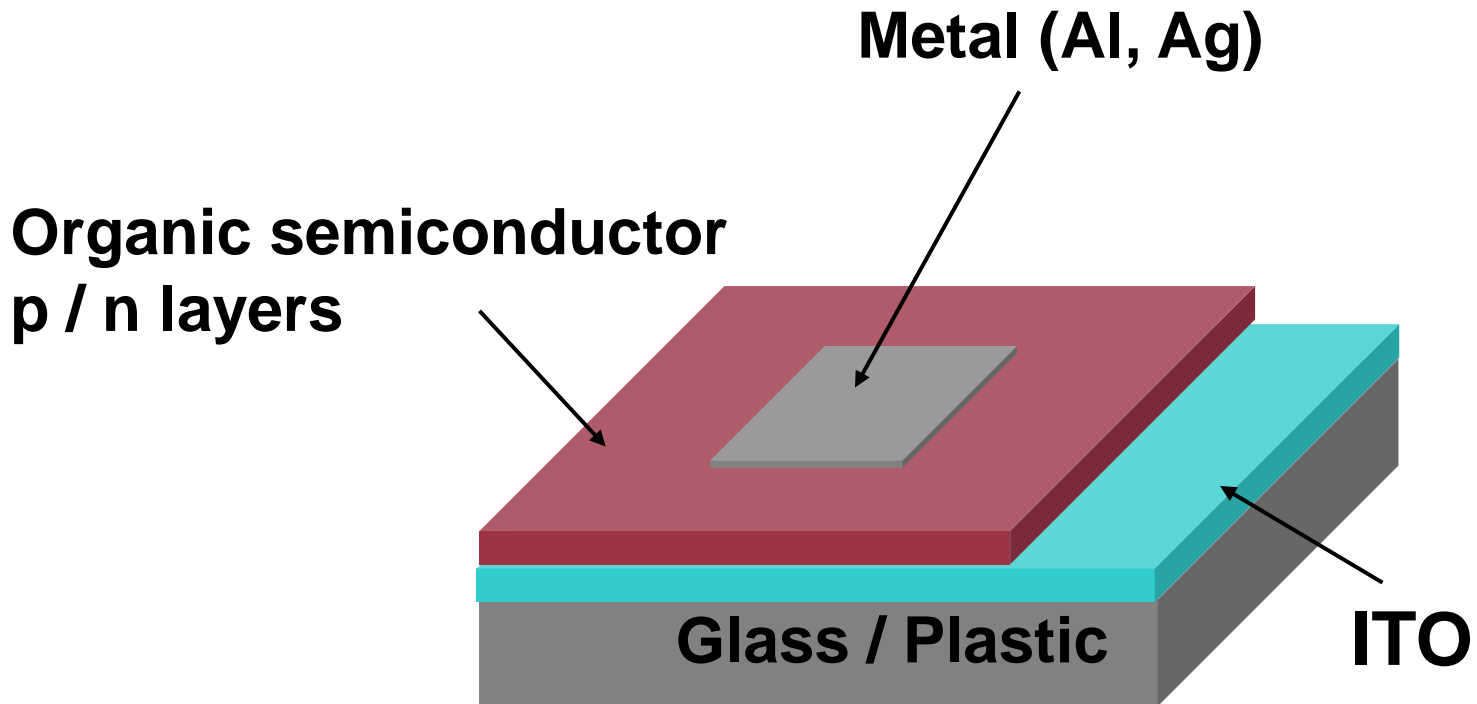
Polymer

Solution
spin coating

Small molecule

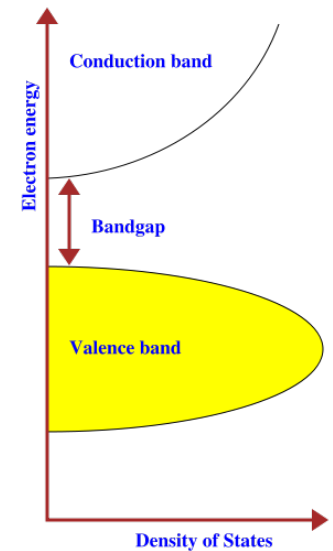
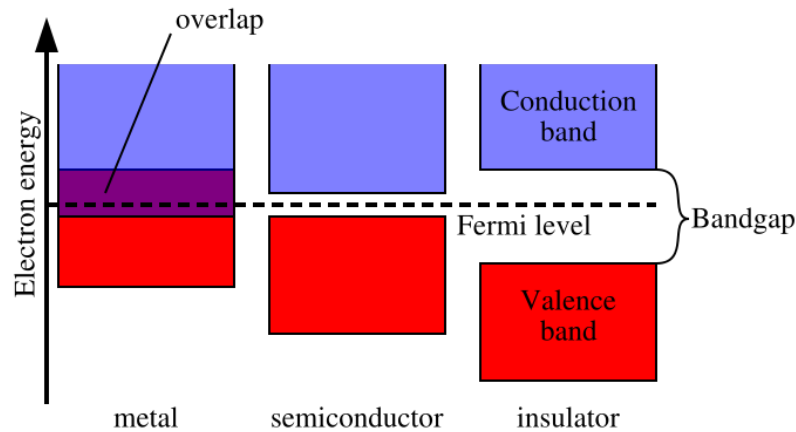
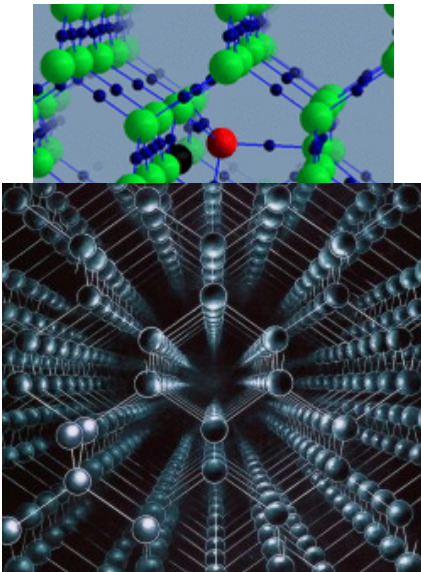
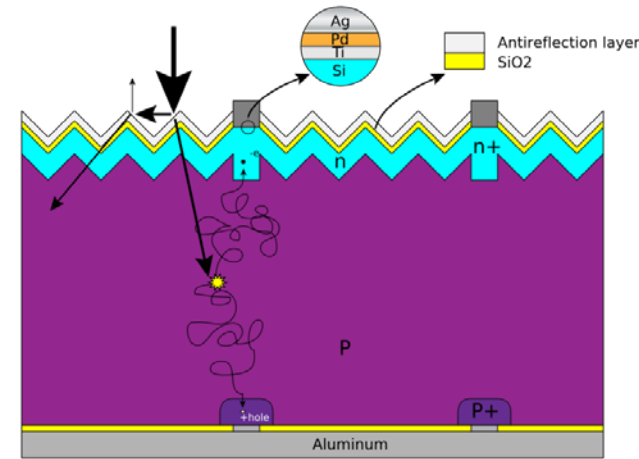
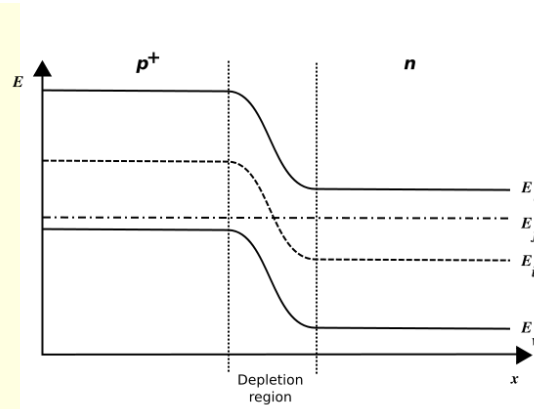
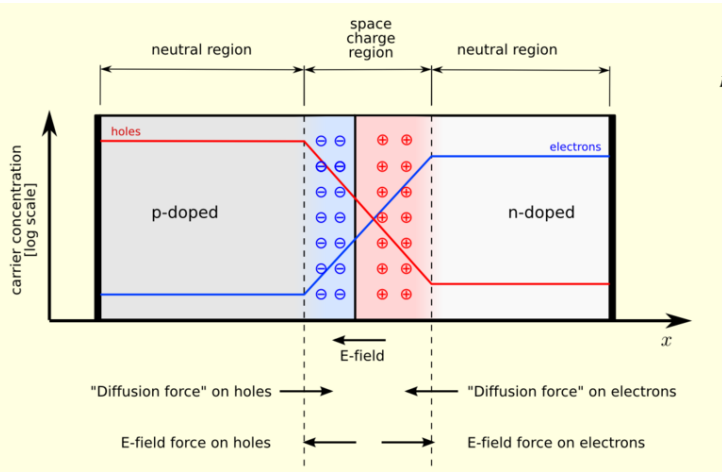
Thermal evaporation
(sublimation) in high-vacuum

Basic structure



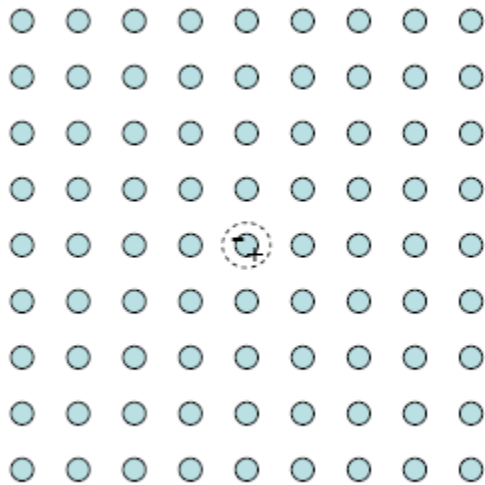
ITO: Indium Tin Oxide / Conductor and transparent
Usually deposited by Sputtering

Inorganic Solar cell (Crystalline silicon)



Exciton

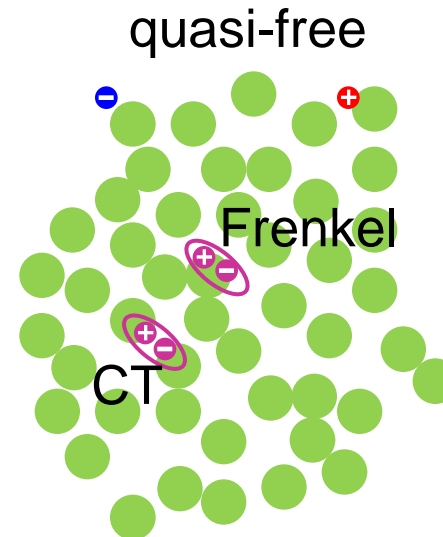
Exciton: electron - hole pair
(molecular or Frenkel exciton)



Frenkel exciton

$$F = - e^2 / 4 \pi \epsilon \epsilon_0 r^2$$

Large binding energy
($\gg kT$) due to the low
dielectric constant

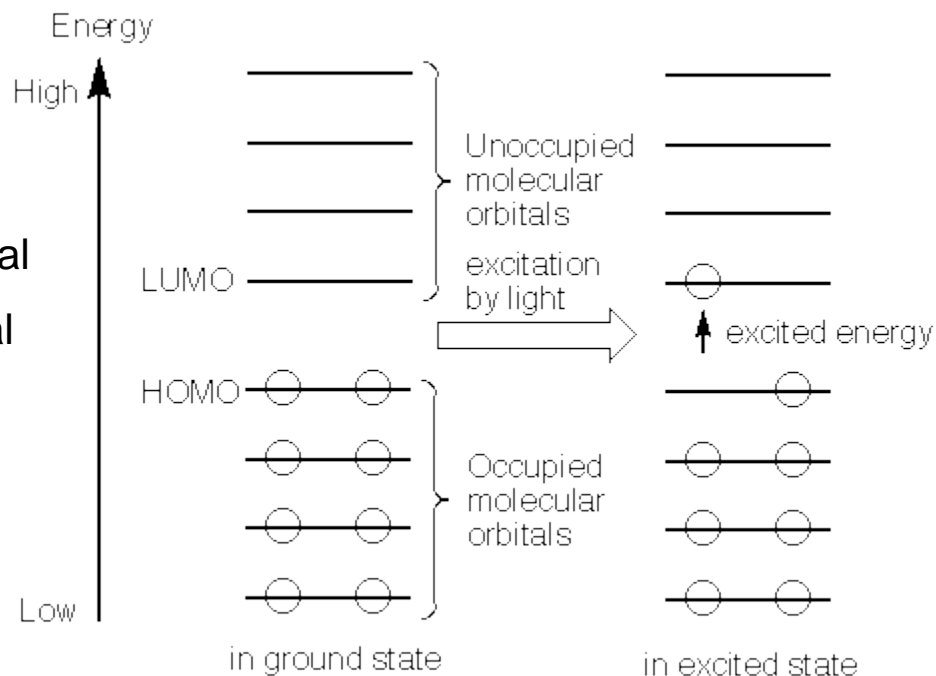


- Quasi-free charge carriers
- Frenkel exciton
- Charge-transfer (CT) exciton

HOMO LUMO levels

LUMO Low Unoccupied Molecular Orbital

HOMO High Occupied Molecular Orbital



LUMO similar to conduction band

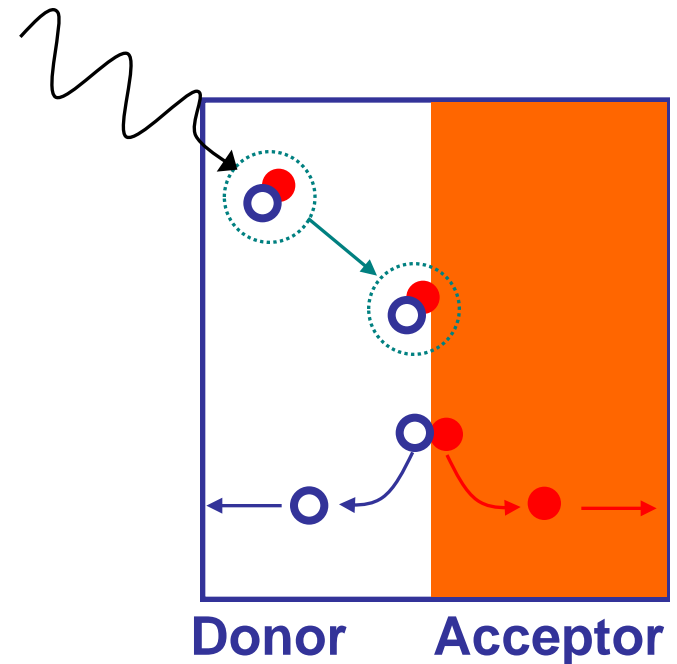
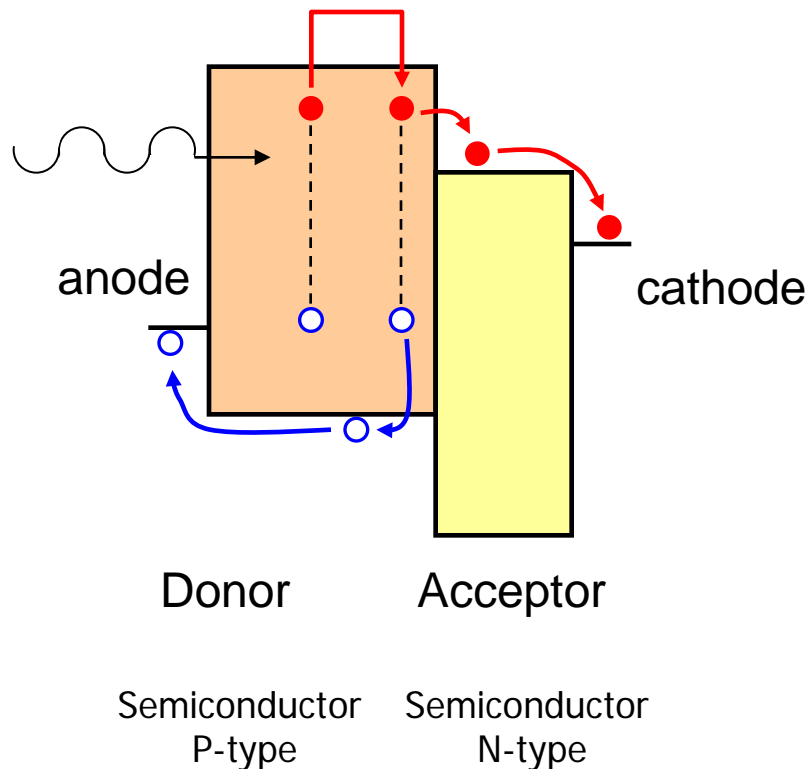
HOMO similar to valence band

IP Ionization Potential [remove electron]

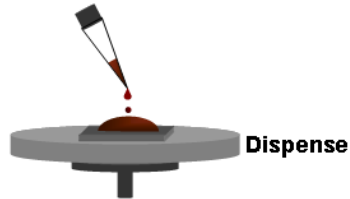
EA Electron affinity [energy gained when an electron is added].

Photocurrent generation

- 1 Photon absorption and exciton generation
- 2 Exciton diffusion
- 3 Charge Transfer
- 4 Carrier collection

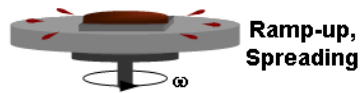


Polymer solar cell

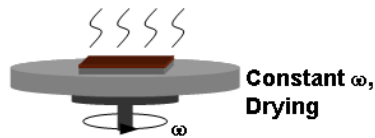


Dispense

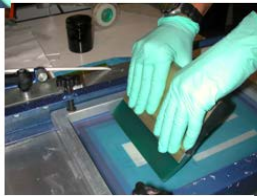
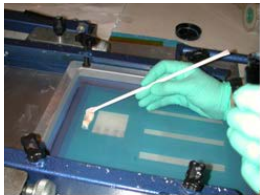
Spin coating



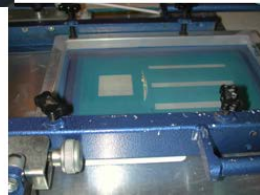
Ramp-up, Spreading



Constant ω , Drying



Dr Blade technique



Semiconductor



Dissolution
(different solvent)



Liquid distribution
(spin-coating, or
Dr Blade technique)

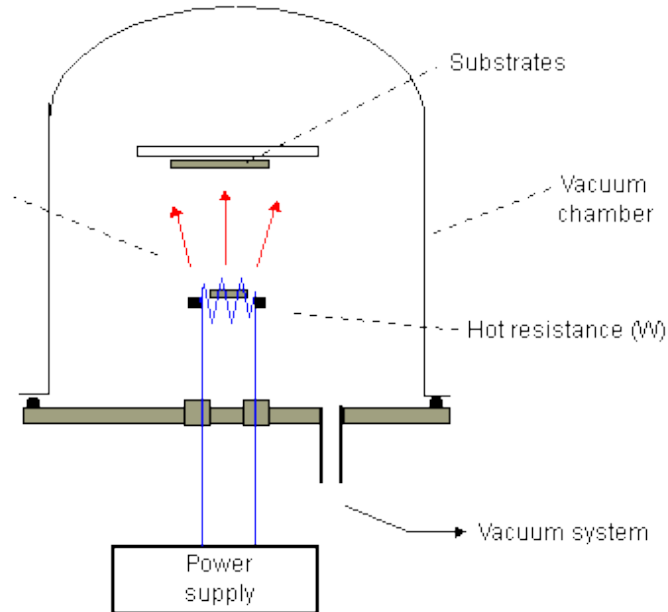
Small molecule solar cell



Semiconductor (powder)



Thermal evaporation in high vacuum



GloveBox



**Our research activities
at UPC on Organic
Solar cells**

Evolution



Year 2002

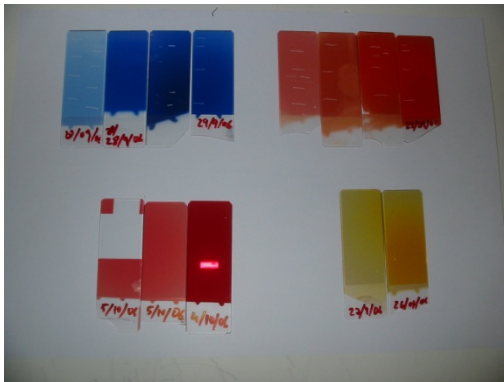
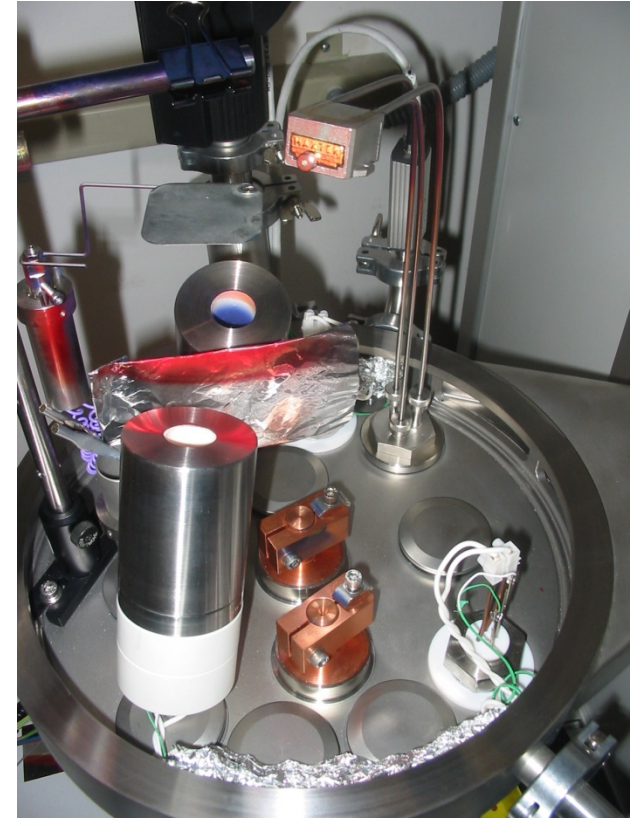


Year 2006

Year 2011

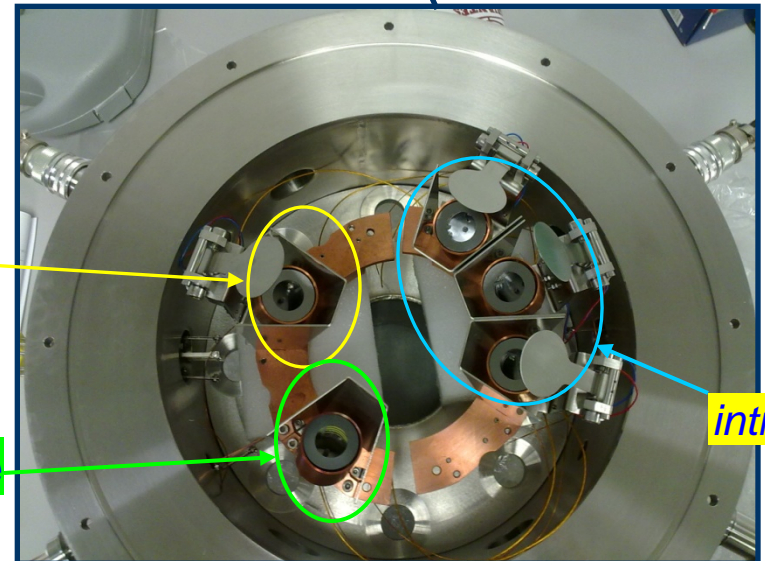


Thermal evaporation





Metal evaporation



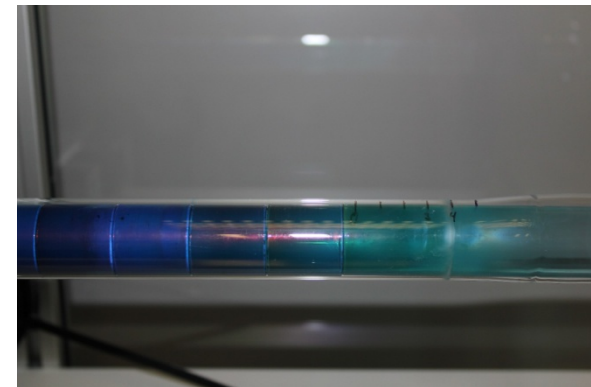
p-type

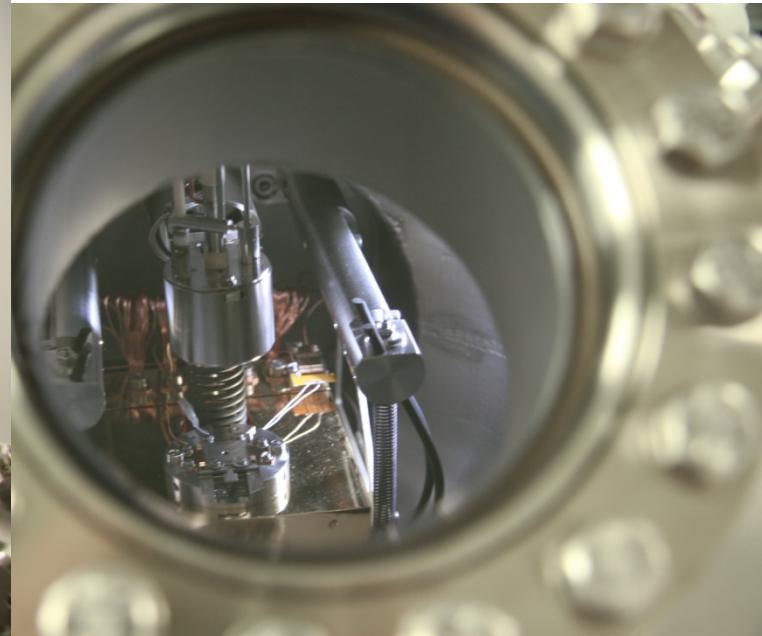
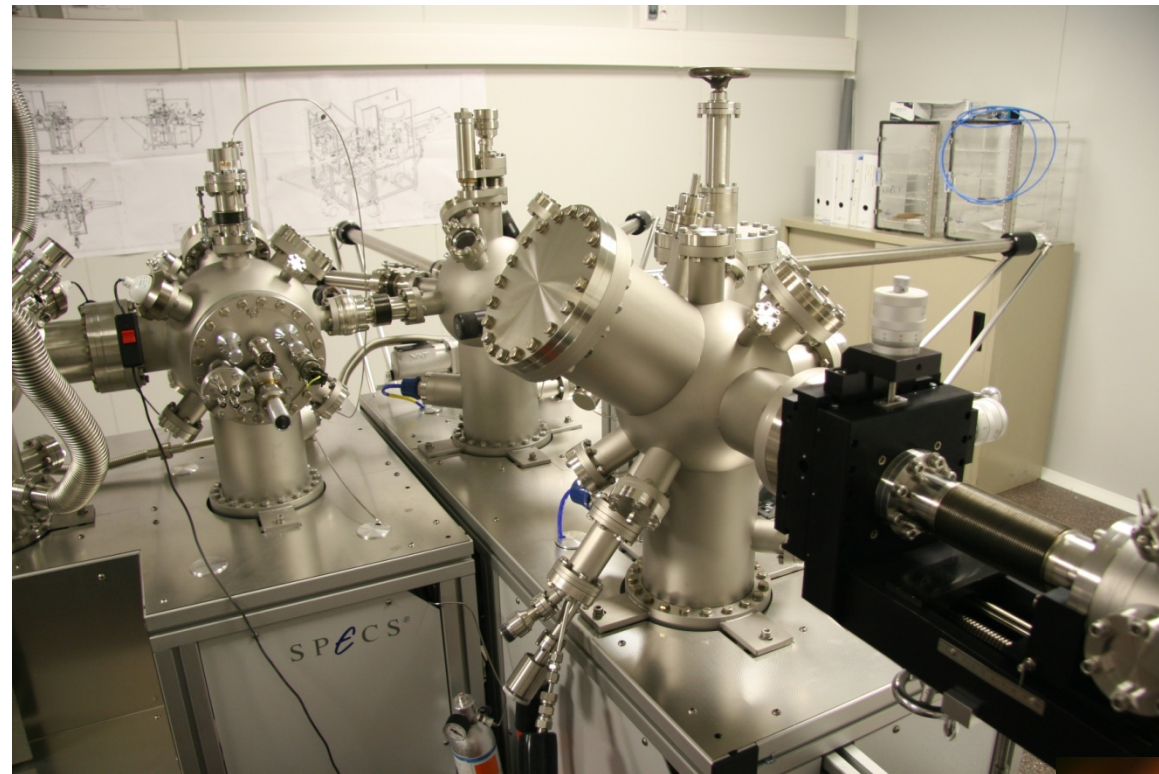
n-type

intrinsic

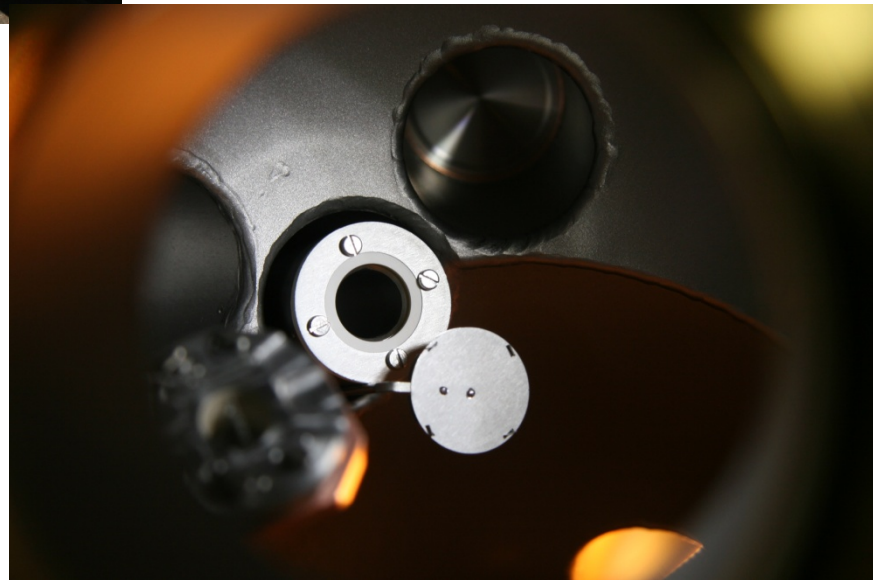
Organic evaporation

Organic semiconductor purification by gradient thermal sublimation

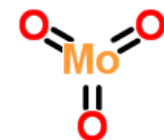
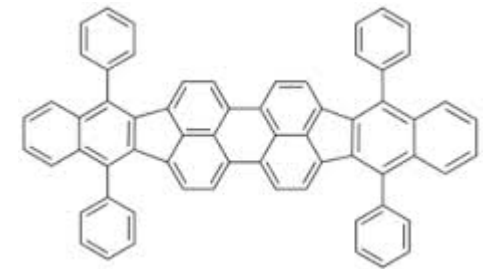
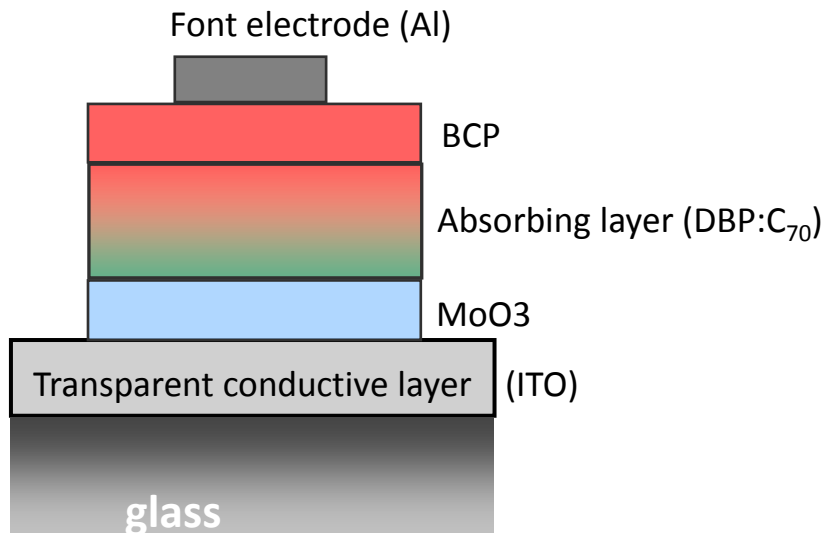
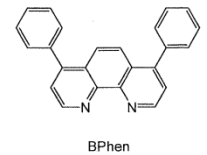
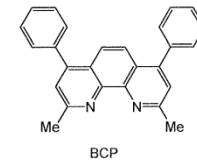
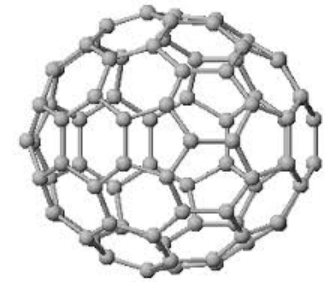
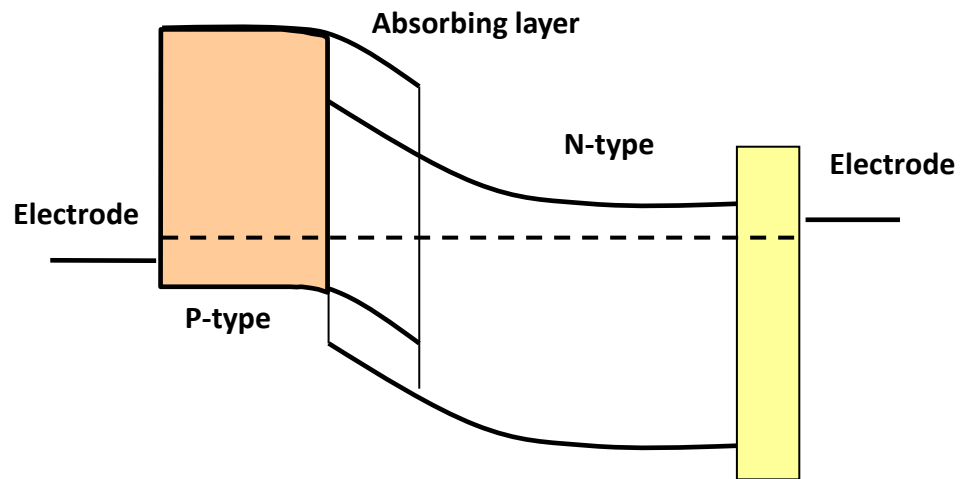




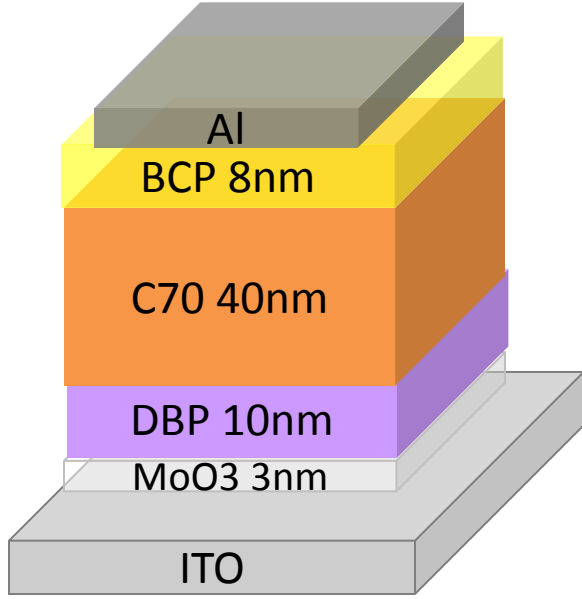
AFM and STM microscope UHV.
Small-molecule thermal deposition



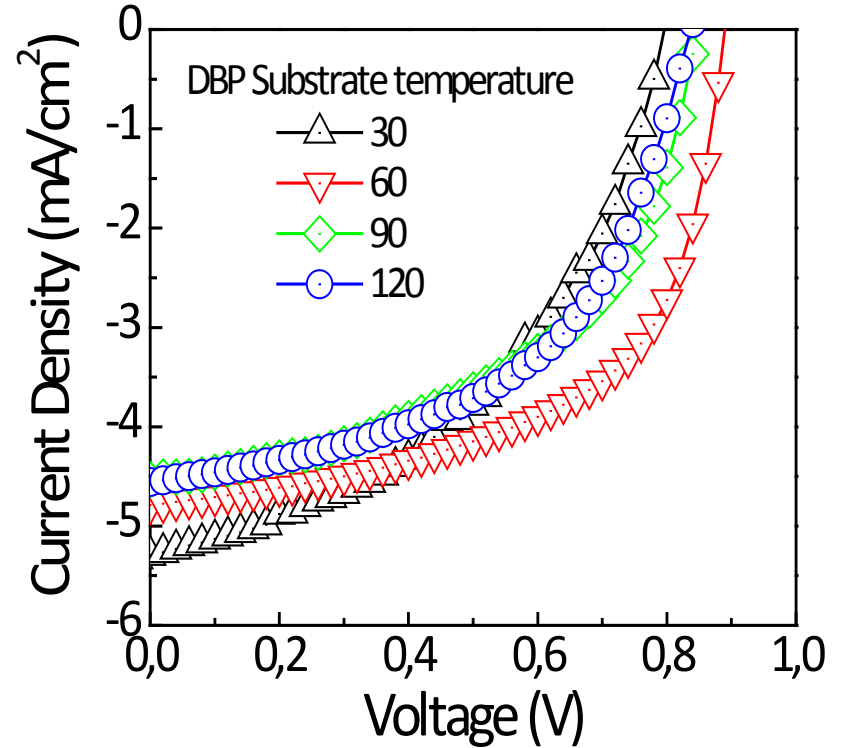
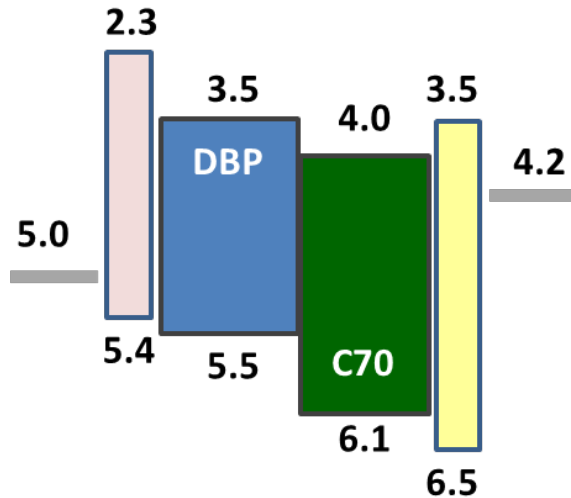
Our solar cell p-i-n



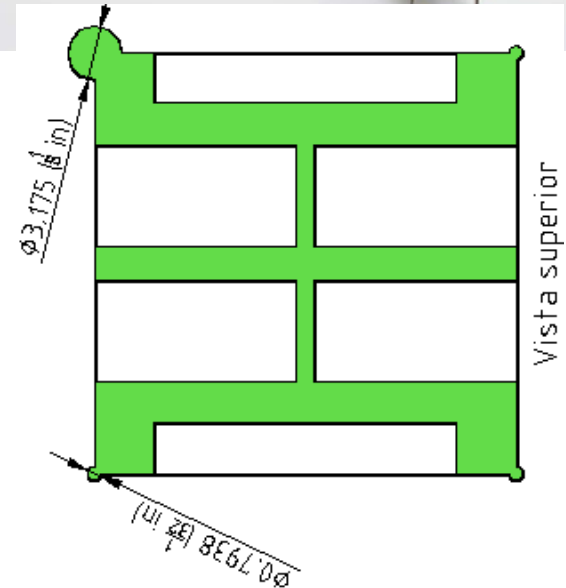
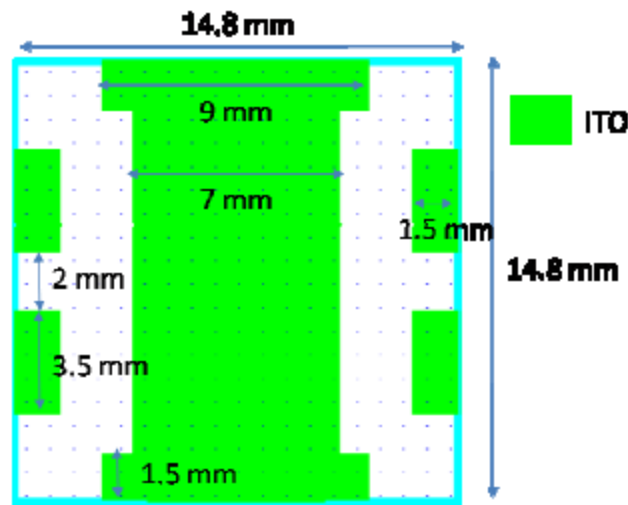
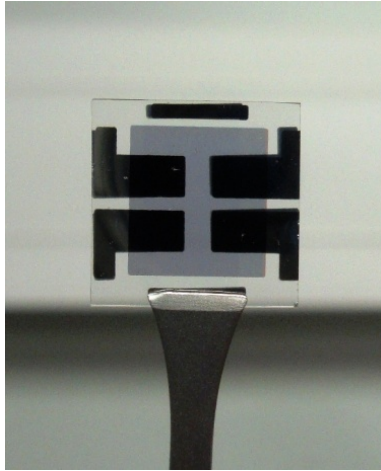
Bilayer solar cell



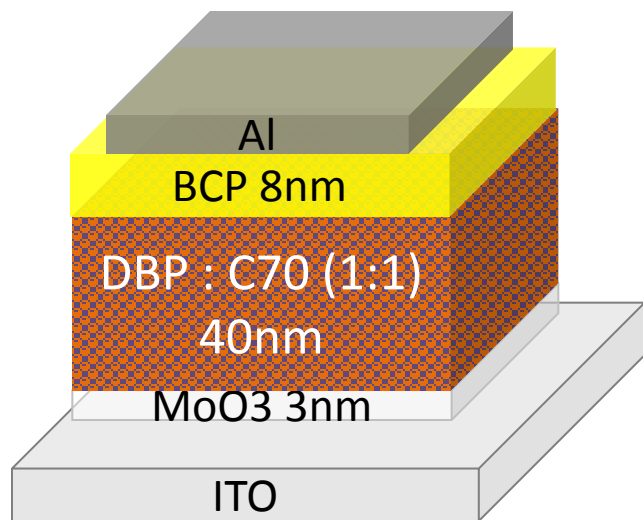
ITO MoO₃ DBP C70 BCP Al



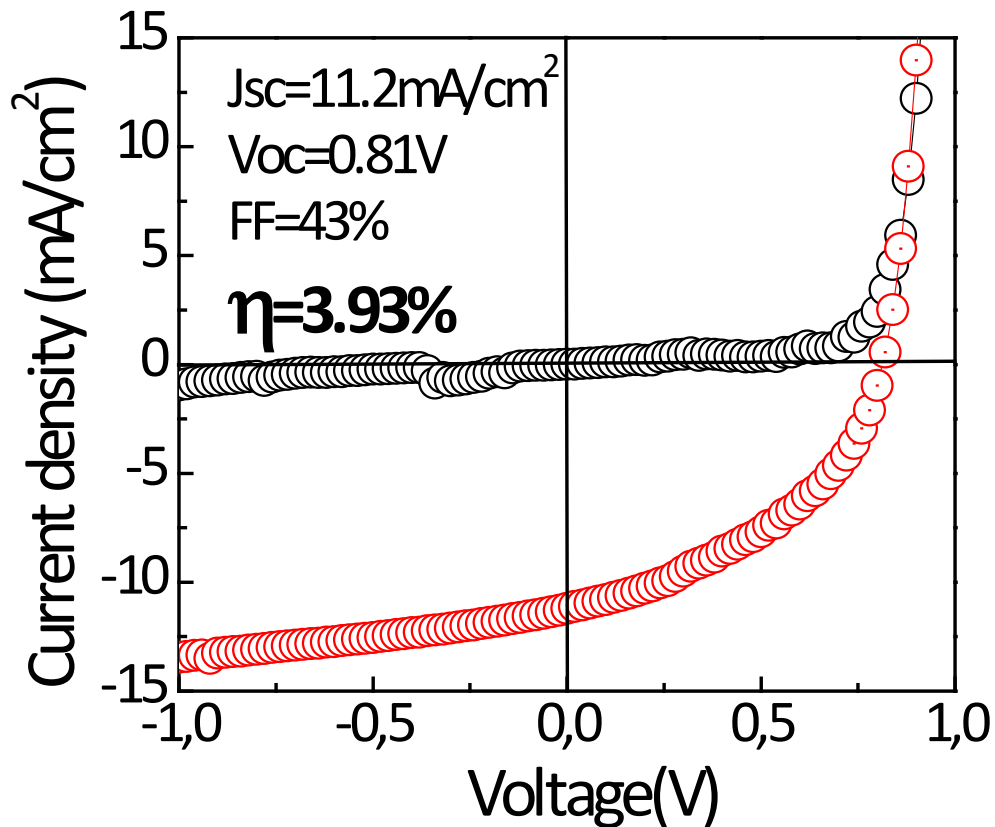
| Temp (C) | PCE (%) | Voc (V) | J _{sc} (mA/cm ²) | FF |
|----------|---------|---------|---------------------------------------|------|
| 30 | 1.92 | 0.79 | -5.29 | 0.46 |
| 60 | 2.48 | 0.89 | -4.79 | 0.58 |
| 90 | 1.96 | 0.85 | -4.53 | 0.51 |
| 120 | 1.98 | 0.83 | -4.56 | 0.52 |



Coevaporated solar cell



$T_{\text{SUBS}} = 60\text{ }^{\circ}\text{C}$
Coevaporated $\sim 4\%$

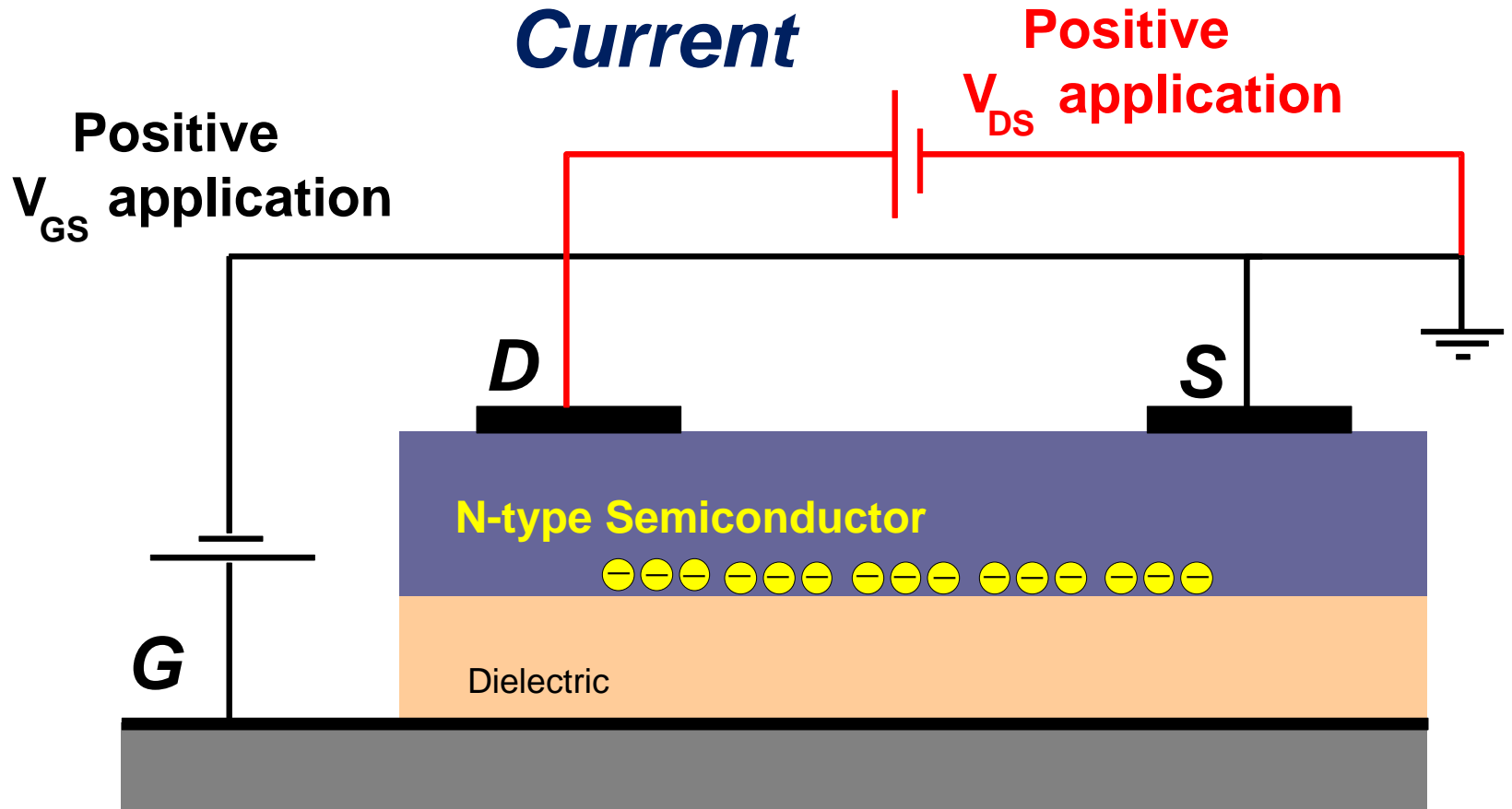


Macko J.A., Lunt R.R., Osedach T.P., Brown P.R., Barr M.C., Gleason K.K., Bulovic V., Phys. Chem. Chem. Phys. 14, 14548–14553 (2012)

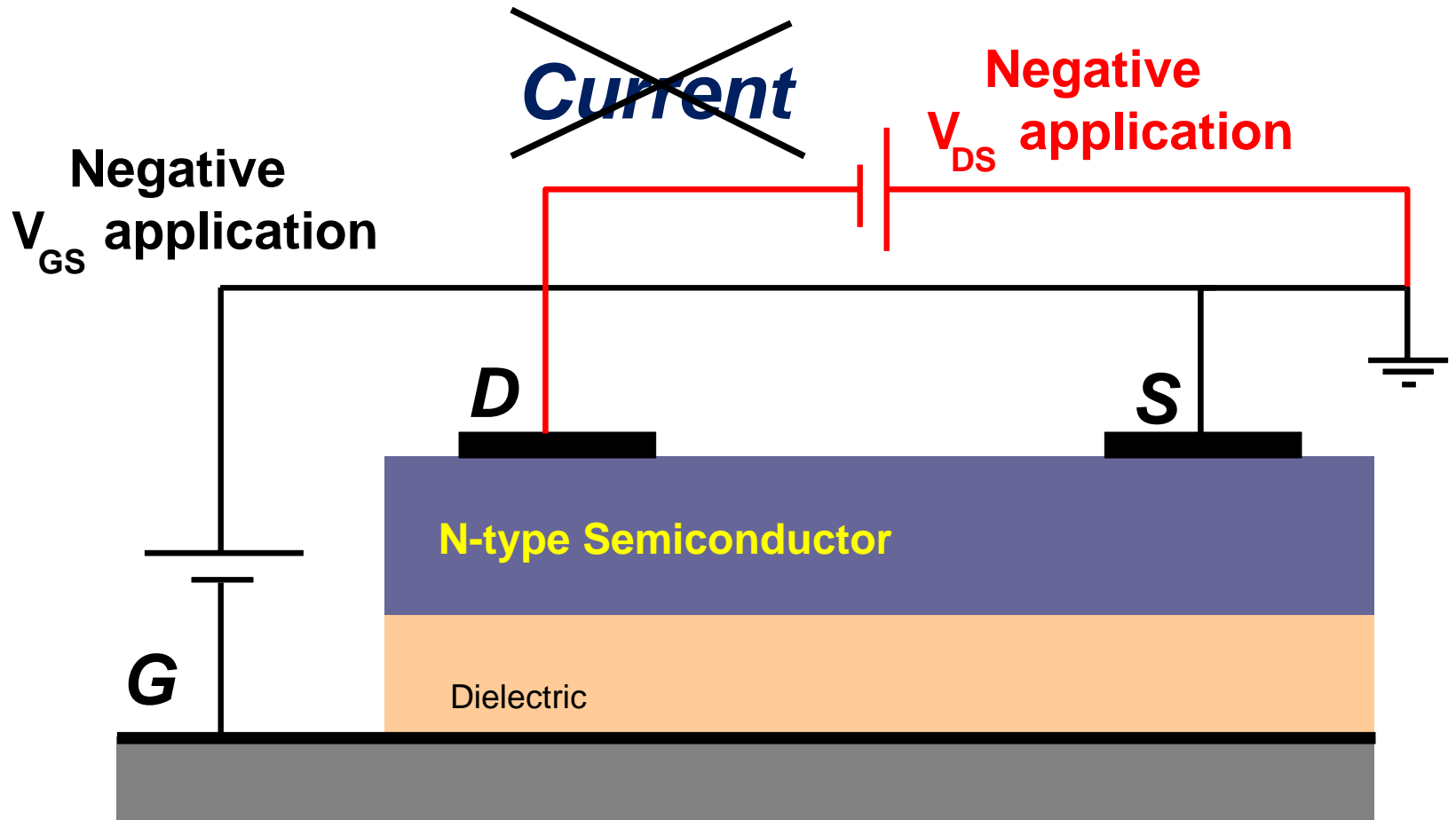
X. Xiao, J. D. Zimmerman, B. E. Lassiter, K. J. Bergemann, S. R. Forrest, Appl. Phys. Lett. 102, 073302 (2013)

Organic Thin-Film Transistors (OTFTs)

Working principle TFTs

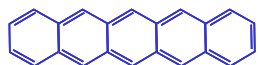


Working principle TFTs

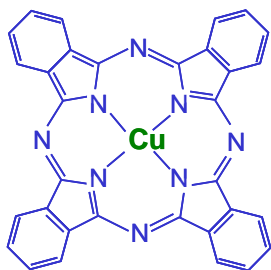


OTFTs allows to determine field-effect mobility (μ)

- μ is an important parameter in organic solar cells
- OTFTs allow to optimize technological parameters



P-type



pentacene

CuPc

Carbazole

Picene

TTF-TCNQ

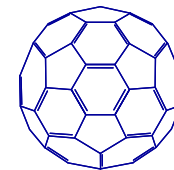
N-type

fullerene (C_{60})

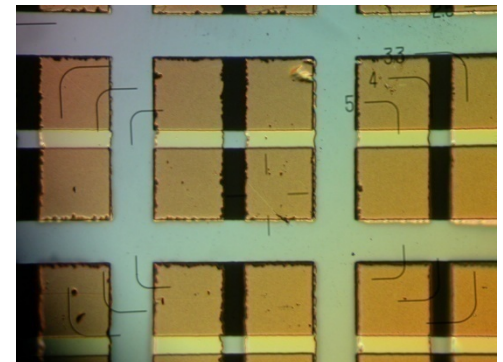
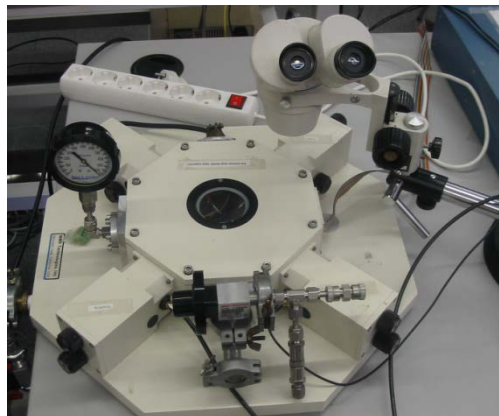
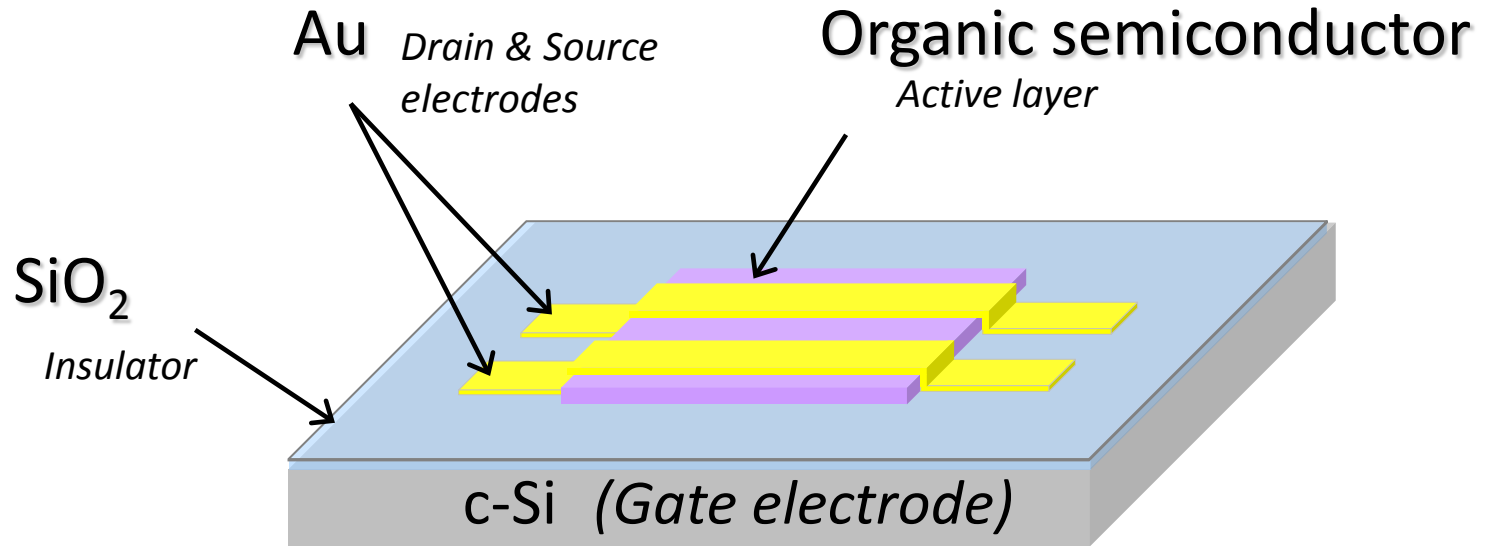
DP-PTCDI

F16CuPc

PTCDI- C_{13}

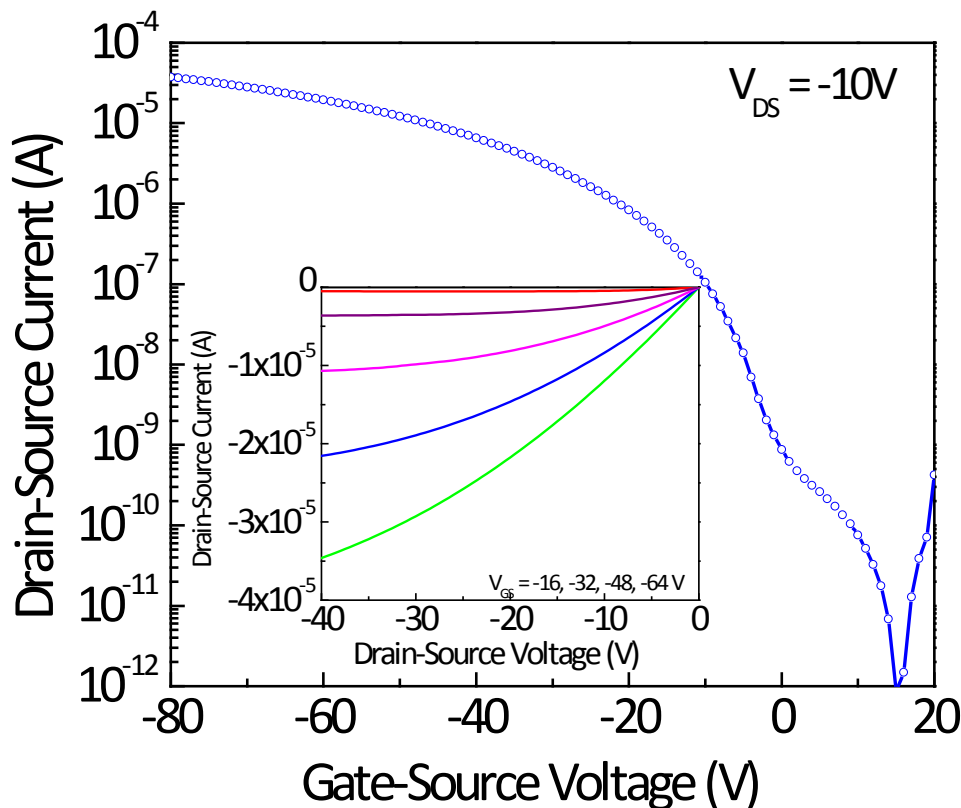


TFTs Structure



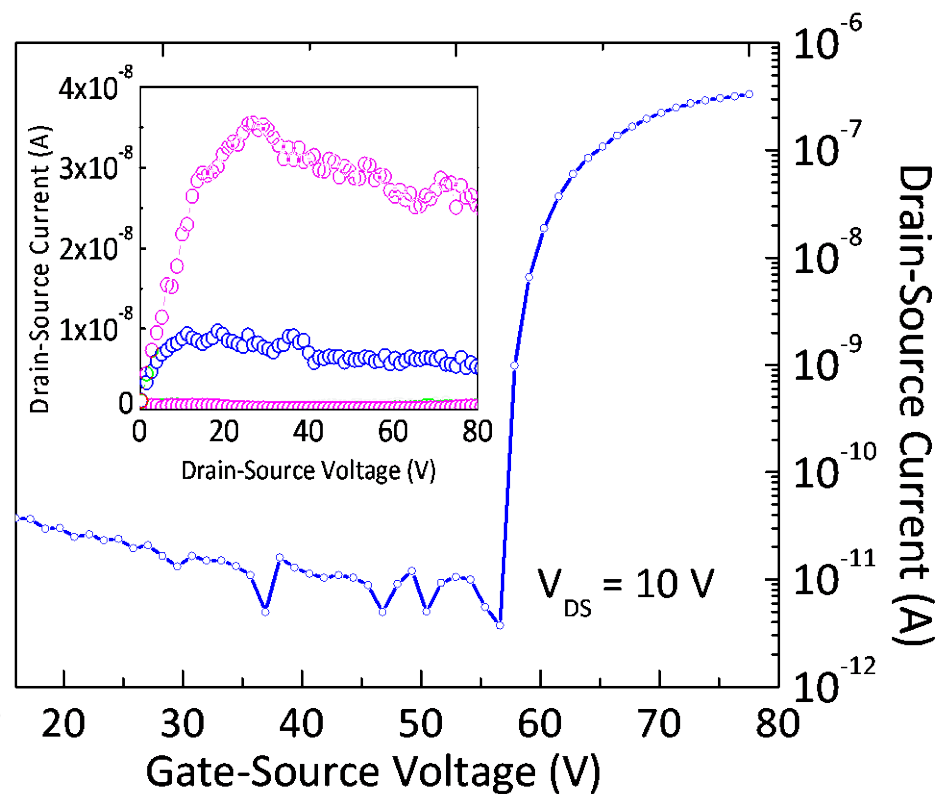
Individual TFT characteristics

Pentacene



$$\mu = 0.5 \text{ cm}^2/\text{V}\cdot\text{s}$$
$$V_T = 15.6 \text{ V}$$

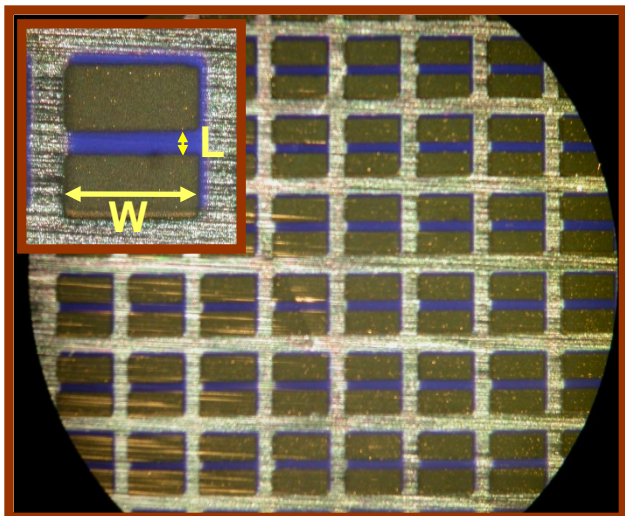
PTCDI-C₁₃



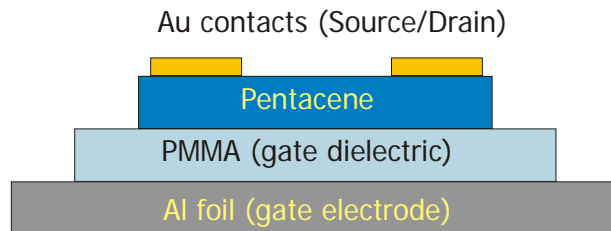
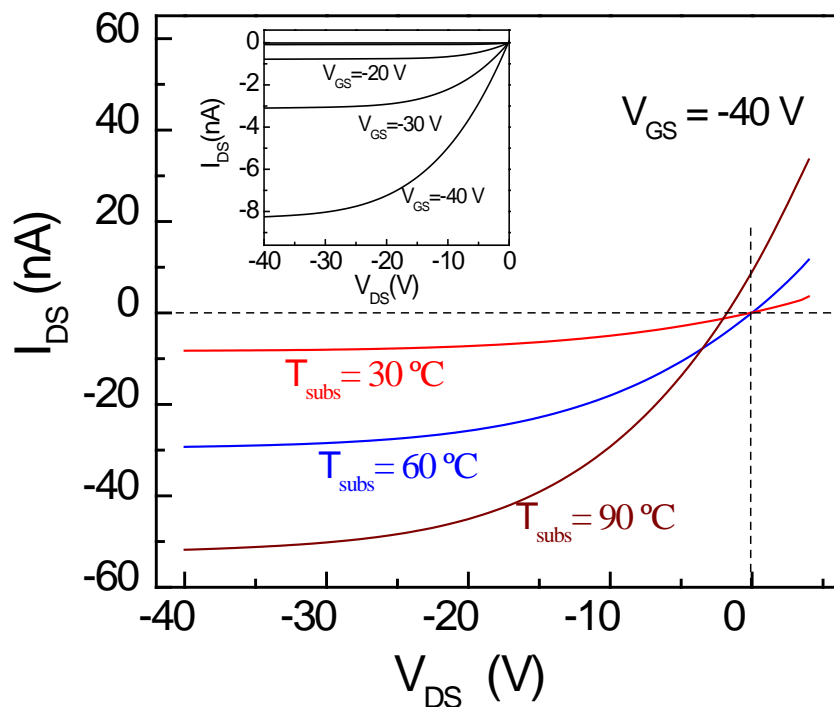
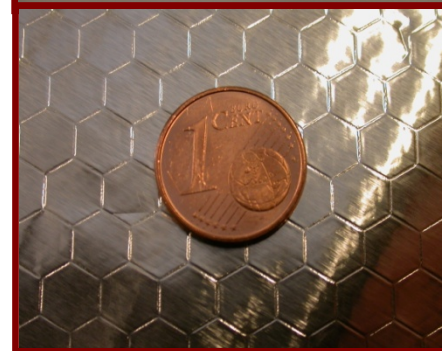
$$\mu = 0.036 \text{ cm}^2/\text{V}\cdot\text{s}$$
$$V_T = 61.7 \text{ V}$$



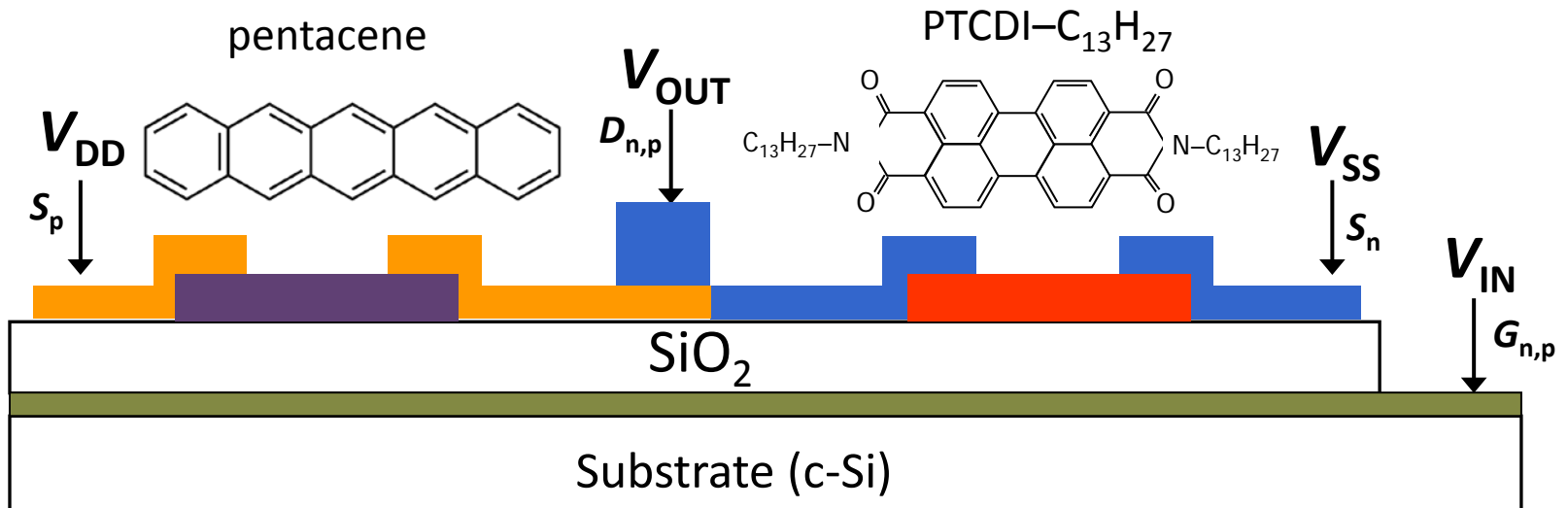
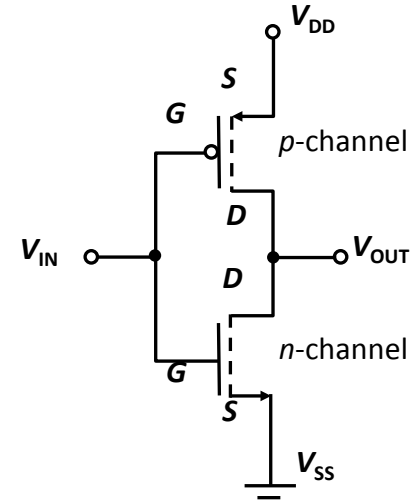
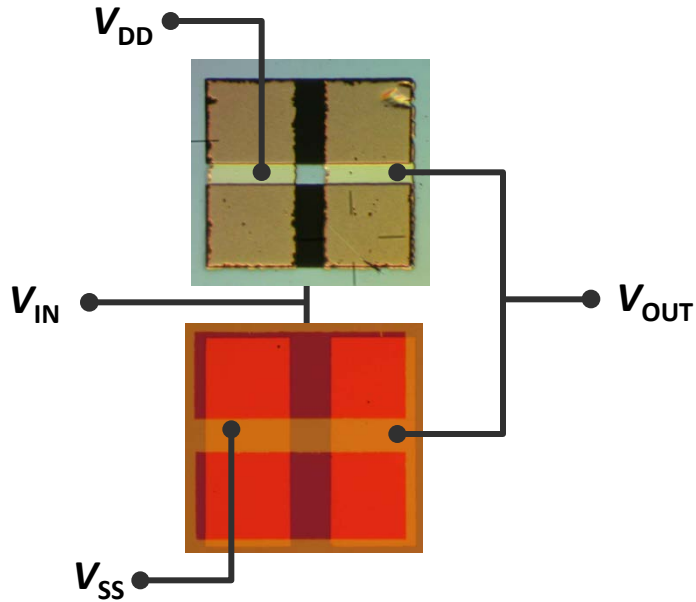
OTFTs on aluminium foil



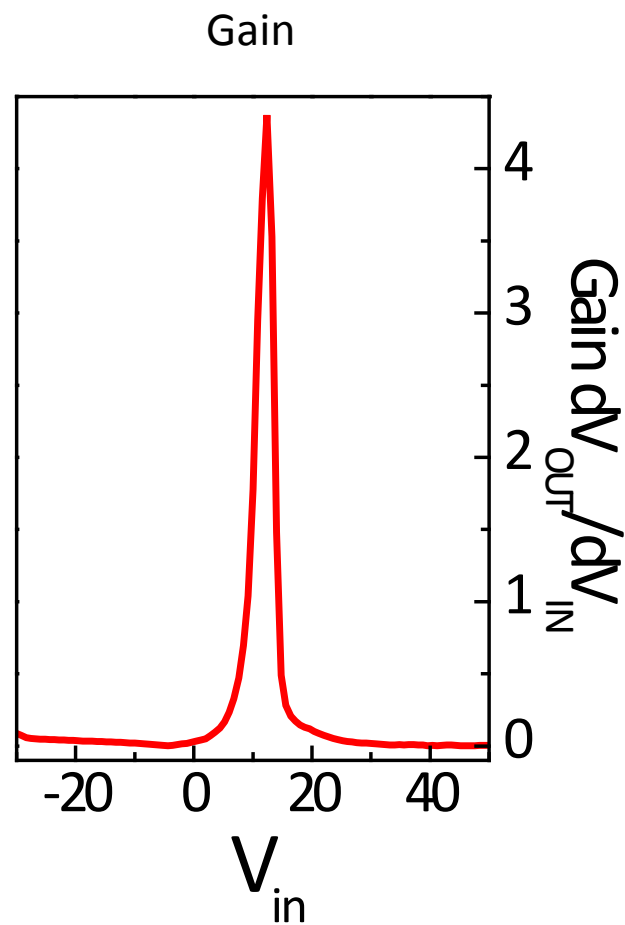
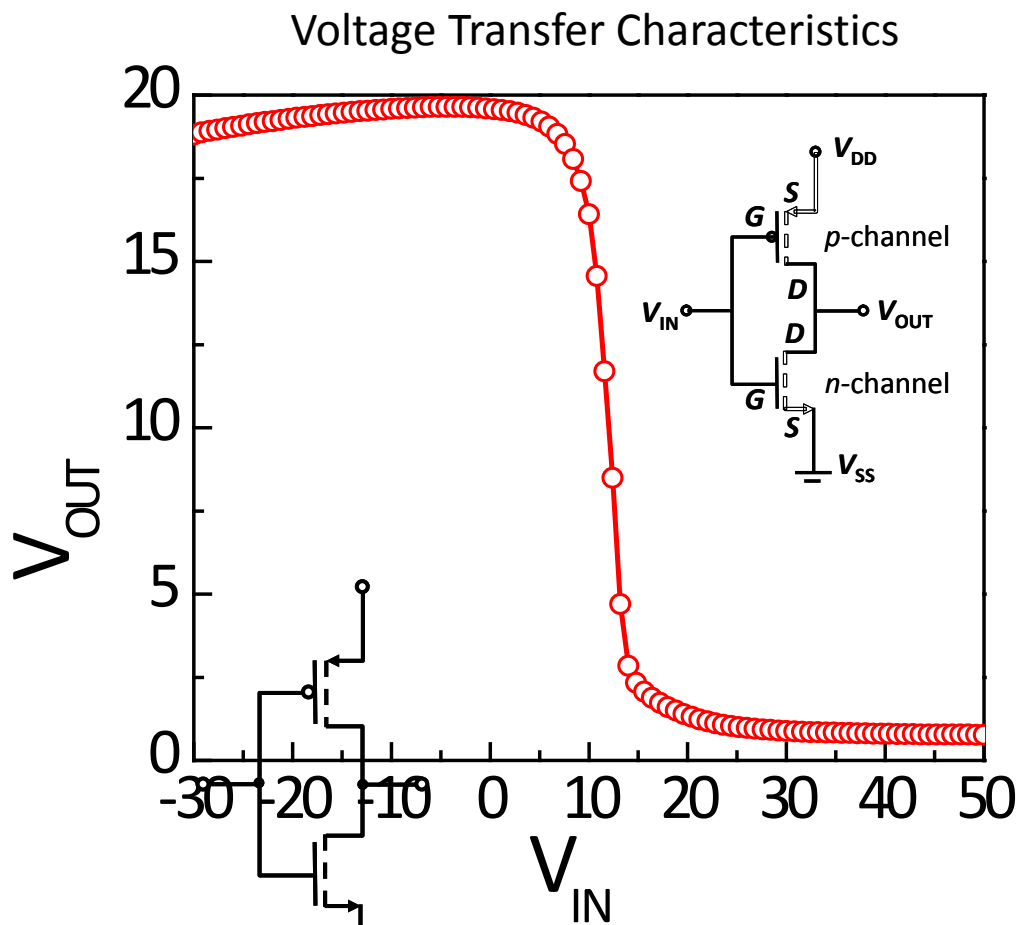
Dielectric PMMA



Complementary Inverter

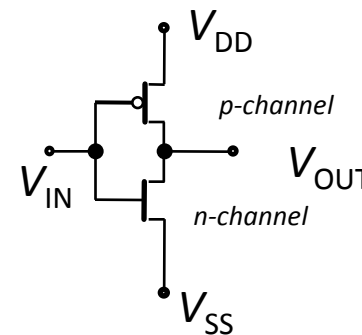
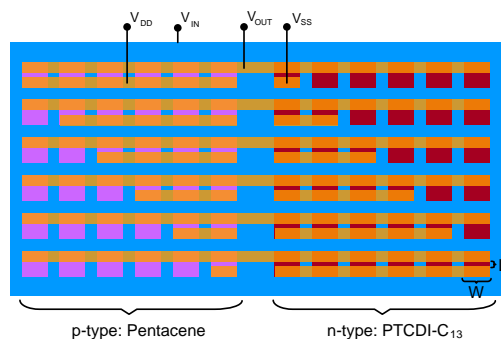
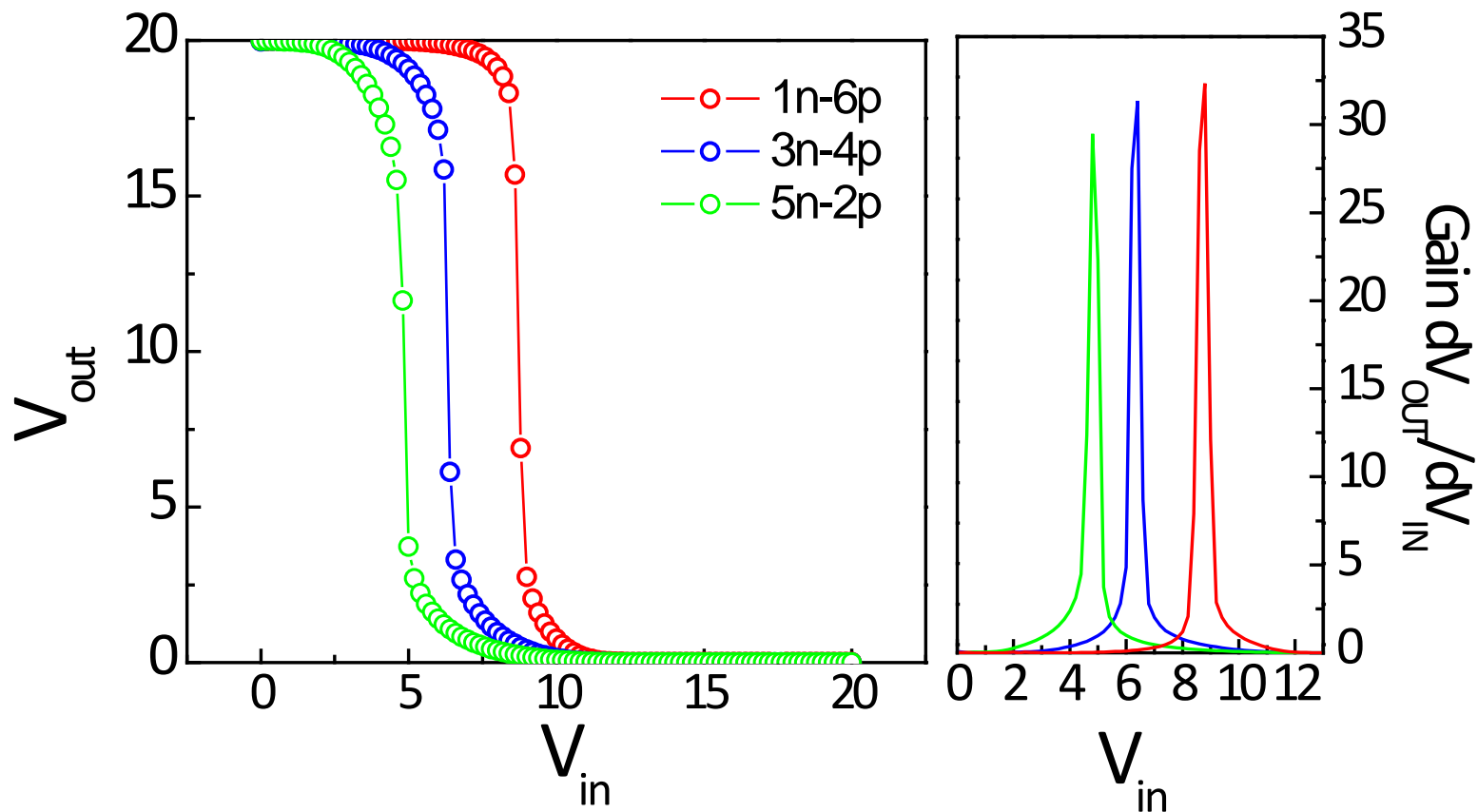


Voltage transfer characteristics



Difficulty to fabricate inverters with symmetrical characteristics

Complementary organic inverters (different W/L)



Thank you